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Powders: Surface Area, Pore Size and Density

Characterization of Porous Solids and Powders: Surface Area and Pore Size

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their behaviour, e.g. chemical reactivity, adsorptive capacity, and catalytic activity is dependent on their pore structure.

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Porous solids are classified as spongy and
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corpuscular. Many corpuscular systems are unconsolidated or only weakly aggregated. If the area of contact between assemblages of globular particles is small, the system will behave in some ways as a non-porous powder.

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All those concerned with the characterization of porous solids in academic and industrial laboratories will find much to interest them in this volume. It should be on the bookshelf of applied research centres involved in adsorption, catalysis, purification of gases and liquids, pigments, fillers, building materials, etc.

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theoretical principles and methodology and (2) to generate proposals for the comparison and utilization of the many techniques now available for the characterization of porous solids.

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work of reference for anyone interested in characterizing porous solids, such as MCM-41 mesoporous materials, pillared clays, etc.

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surfaces - Factors

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Idealization: cube ? $S = 6 l^2$. sphere ? $S = 4 \pi r^2$.

Reality: Surface

roughness due to voids, pores, steps, other imperfections and

surface atomic and

molecular orbitals S_{TM}^{real}

surface area $>$

corresponding

theoretical area •

Particle size: e.g. cube l

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$= 1 \text{ m}^3 \text{ S} = 6 \text{ m}^2$.

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