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□ No two electrons in  
an atom can have the  
same unique set of  
four quantum  
numbers. □ The first  
quantum number  
describes the

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principal energy level.

For example, the quantum number 2 identifies the second principal energy level.

□ The first two quantum numbers together describe a sublevel.

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Character of the

Electron Other

Important Waveforms

Overall Organization

of Principal Energy

Levels, Sublevels,

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... it helped lay the foundation for the modern quantum theory. d.) it helps to locate a proton in an atom. d.) it is the same as the Bohr's theory ... the atomic

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nucleus b.) atomic  
orbitals c.) atoms d.)  
molecules.

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foundation for the modern quantum theory. d.) it helps to locate a proton in an atom. d.) it helps to locate a proton in an atom. ... the atomic nucleus b.) atomic orbitals c.) atoms d.) molecules.

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work. kinetic energy.  
the process of burning  
something that  
creates energy. the  
capacity to do work.  
what is done to move  
an object against  
some sort of  
resistance. the  
energy of motion ( $\frac{1}{2}mv^2$ )

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Chapter 4. List the three statements that make up the modern atomic theory. 2. Define atomic number. What is the atomic number for a boron atom? 3. Which elements are alkali metals, alkaline earth metals, and

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halogens? 4. Two of the most well-known isotopes of carbon are carbon-12 and carbon-14.

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Objective 3. 11.1 The

Mysterious Electron

417 the electron are also called orbitals.

The orbital shown in Figure 11.3 is called the 1s orbital. The

negative charge is most intense at the nucleus and

decreases in intensity with distance outward.

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Chapter 4 Atomic

Structure Section 4.3

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Theory (pages

113118). Use this

table to answer

questions 12 and 13..

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Worksheet. Answer

the following

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questions, circle the  
best answer.

Chapter 4 Atomic  
Structure Worksheet  
Answer Key Pdf

The modern atomic  
theory states that  
atoms of one element  
are the same, while  
atoms of different  
elements are  
different. What makes  
atoms of different

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elements different?

The fundamental characteristic that all atoms of the same element share is the number of protons. All atoms of hydrogen have one and only one proton in the nucleus; all atoms of iron have 26 protons in the nucleus.

Atomic Theory □

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separate atoms

(Figure 4.1). Radiant

energy from the sun

causes sunburn 4.1

Energy Some

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reactions absorb

energy. Energy from  
the sun  $O_2(g) +$

energy  $2O(g)$  Figure

4.1 Some reactions  
absorb energy. 120

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we can begin to  
explain the role that  
energy plays in these  
and other chemical

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