

Acids Bases And Salts Worksheet Answer Key

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India. CLASS: X CHEMISTRY. FIRST TERM: ASSIGNMENT-2 (2010) ACIDS, BASES AND SALTS. 1. What are the symptoms of the problem of acidity? What possible remedy can be used at home? 2. Name three chemical indicators.

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Acid Base and Salt Worksheet Chapter 5. In this page we have Worksheet for Class 7 Science Chapter 5: Acid Base and Salt. Hope you like them and do not forget to like , social share and comment at the end of the page. > Question 1 Fill in the blanks (a) The most commonly used natural indicator is _____ and is extracted from _____ ...

~~Worksheet for Class 7 Science Chapter 5: Acid Base and Salt~~

Home > Science Worksheets > Acid and Base. Liquids are classified as acids, bases, or neutral based on the concentration of hydrogen ions present in it. If a substance is neutral it has an equal amount of positive and negative ions present in it. If a liquid has a greater concentration of positive ions in the form of hydrogen than say water, it is classified as an acid.

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Acids, Bases, and Salts . Properties of Acids and Bases: This cloze activity reviews the basic properties of acids and bases. The pH Scale: This cloze covers the pH scale for measuring the strength of acids and bases. Acids, Bases, and Salts: The cloze covers vocabulary and terms related to acid-base neutralization.

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Acids, bases and salts unit. 4.8 15 customer reviews. Author: Created by Masfar. Preview. Created: Feb 20, 2012 | Updated: May 9, 2013. Credit to eucalyptusoliver for giving the outline of the booklet which I adapted for use at IGCSE. Read more. Free. Loading... Save for later.

~~Acids, bases and salts unit | Teaching Resources~~

1. Nitric acid turns red litmus blue. ☐☐☐☐.. 2. Sodium hydroxide turns blue litmus red. ☐☐☐☐☐. 3. Sodium hydroxide and hydrochloric acid neutralize each other and form salt and water. ☐☐☐☐☐. 4. Indicator is a substance which shows different colours in acidic and basic solutions. ☐☐☐☐☐. 5.

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1. acids 2. citric acid 3. Base 4. calcium sulphate hemihydrate, $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$ 5. yellow colour, pink colour 6. Hydrochloric Acid (HCl) 7. Acetic Acid (CH_3COOH) 8. Sodium Hydroxide (NaOH) 9. Ammonium Hydroxide (NH_4OH) 10. sodium carbonate, $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$, permanent hardness 11. PH 12. Acid 13. Base 14. Neutral 15. decreases 16. Salt and Water 17. Calcium Carbonate 18. 5.5

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